## Production of purified <sup>85</sup>Sr solution

S. Yano,<sup>\*1,\*2</sup> Y. Wakitani,<sup>\*1,\*2</sup> T. Yamada,<sup>\*1,\*2</sup> J. Kanaya,<sup>\*2</sup> S. Shibata,<sup>\*2</sup> and H. Haba<sup>\*2</sup>

Since 2007, we have distributed purified radioisotopes such as  ${}^{65}$ Zn,  ${}^{88}$ Y, and  ${}^{109}$ Cd to the general public.<sup>1)</sup> After the Fukushima Dai-ichi Nuclear Power Plant accident in 2011, the demand for  ${}^{85}$ Sr solutions having a high specific radioactivity has been growing. In this work, we investigated the production of  ${}^{85}$ Sr in the  ${}^{85}$ Rb(d,2n) ${}^{85}$ Sr reaction using a 24-MeV deuteron beam from the RIKEN AVF cyclotron. We also studied a chemical procedure to obtain a purified  ${}^{85}$ Sr solution.

<sup>85</sup>Sr ( $T_{1/2}$  = 64.853 d) was produced by irradiating an RbCl disk (Sigma-Aldrich; chemical purity: > 99.99%; thickness: 500 mg cm<sup>-2</sup>) of natural isotopic abundance with 24-MeV deuterons. The average beam intensity was 159 nA. The irradiation time was 17 min. 85 Sr was chemically separated in accordance with the scheme shown in Fig. 1. The irradiated RbCl target was dissolved in 2 mL of H<sub>2</sub>O and 2 mL of 8 M HNO<sub>3</sub>. After evaporating the solution almost to dryness, the residue was again dissolved in 2 mL of 8 M HNO<sub>3</sub>. The resulting solution was evaporated to dryness to remove chloride ions, and the residue was dissolved in 4 mL of 8 M HNO<sub>3</sub> and loaded onto a reversed-phase extraction chromatography column ( $\phi$  5 mm×50 mm height) packed with Sr Resin (Eichrom; 100-150 mesh). The column was then washed with 12 mL of 8 M HNO<sub>3</sub>. In this process, 85Sr was absorbed on the Sr Resin, and the target material of Rb was completely eluted, as traced with byproducts of <sup>84</sup>Rb and <sup>86</sup>Rb. <sup>85</sup>Sr was then eluted with 8 mL of 0.05 M HNO<sub>3</sub>. The eluent was evaporated to dryness, and the residue was dissolved in 1 mL of concentrated HCl (c. HCl). After evaporating the solution almost to dryness, the residue was dissolved in 2 mL of 0.1 M HCl and loaded onto a column ( $\phi$  5 mm×40 mm height) packed with a cation-exchange resin (Dowex 50W×8; 200-400 mesh). The column was then washed with 3 M HCl. <sup>85</sup>Sr was eluted with 6 M HCl. The activity of <sup>85</sup>Sr was determined through  $\gamma$ -ray spectrometry using a calibrated Ge detector.

The  $\gamma$ -ray spectra of the produced <sup>85</sup>Sr are shown in Fig. 2. The produced activity of <sup>85</sup>Sr was 145 kBq, and the radionuclidic purity was > 99.9%. The production yield of <sup>85</sup>Sr under the present experimental condition was about 3 MBq  $\mu$ A<sup>-1</sup> h<sup>-1</sup>. The chemical yield was 89%.

The chemical impurity in the purified solution will be evaluated by using ICP MS for a control sample, which was treated using the same procedure as that used for the irradiated sample, in further studies.



Fig. 1. The chemical separation procedure of <sup>85</sup>Sr from the irradiated RbCl target employed in the present study.



Fig. 2. The measured  $\gamma$ -ray spectra of the produced <sup>85</sup>Sr. (a) and (b) were obtained before and after the chemical separation, respectively.

## References

1) T. Kambara et al.: a separate paper in this issue.

<sup>\*1</sup> Japan Radioisotope Association

<sup>\*2</sup> RIKEN Nishina Center