Na diffusion in $Na_x FeO_2$

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In order to develop next-generation Na-ion batteries,¹⁾ we have attempted to measure Na diffusion in several candidate materials with μ^+ SR.²⁾ The present target compound, α -NaFeO₂, has a layered rocksalt structure with the trigonal space group $R\bar{3}m$ (Fig. 1). The isostructural compound, NaCoO₂, works as a positive electrode material owing to reversible Na⁺ deintercalation from and intercalation into the lattice.³⁾ Furthermore, μ^+ SR clearly detected Na diffusive behavior in Na_xCoO₂ at high T,²⁾ which is consistent with electron diffraction and neutron scattering results.⁴⁾ However, α -NaFeO₂ does not exhibit reversible Na deintercalation and intercalation behavior.⁵⁾

The following two explanations were proposed for such behavior in α -NaFeO₂, 1) Na ions do not diffuse, or 2) Na ions diffuse, but electrons are not transferred for charge compensation. Since α -NaFeO₂ is an antiferromagnetic insulator,^{6,7)} the second explanation is more reasonable.

Figure 2 shows the *T* dependences of the field fluctuation rate (ν) and field distribution width (Δ) for α -NaFeO₂ and Na_{0.7}CoO₂ extracted from the μ^+ SR spectra measured on ARGUS. Here, ν corresponds to the spin-lattice relaxation rate (T_1^{-1}) of NMR, and Δ corresponds to the spin-spin relaxation rate (T_2^{-1}). For Na_{0.7}CoO₂, ν increases with *T* particularly above 325 K, while Δ decreases with *T*. This is a typical motional narrowing behavior due to Na diffusion.

For α -NaFeO₂, both ν and Δ are scattered in the whole T range measured. In particular, there is no ac-



Fig. 1. Crystal structure of α -NaFeO₂ consisting of a twodimensional triangular lattice of FeO₂ layers formed by edge sharing FeO₆ octahedra separated by the Na layer.



Fig. 2. *T* dependences of the field fluctuation rate (ν) and field distribution width (Δ) for α -NaFeO₂ and Na_{0.7}CoO₂.

ceptable reason for the increase in Δ above 300 K. Note that even for stoichiometric $NaFeO_2$, there are interstitial sites between two adjacent Na⁺ ions, to which Na^+ ions jump, as in the case for $LiNiO_2$.⁸⁾ Back to the data analysis, the ZF- and LF-spectra were fitted by a combination of an exponentially relaxing dynamic Kubo-Toyabe signal and a time-independent background signal from the muons stopped in the sample holder. This indicates the presence of two muon sites in the lattice, although the muon sites are unknown at present. However, owing to the large exponential relaxation of the two terms caused by the fluctuation of the Fe³⁺ moments, it is very difficult to extract ν and Δ from the present μ^+ SR spectrum with the statistics of 18M events. Thus, we need additional data with higher statistics.

References

- 1) S-W. Kim et al., Adv. Energy Mater. 2, 710 (2012).
- 2) M. Månsson, J. Sugiyama, Phys. Scr. 88, 068509 (2013).
- L. W. Shacklette et al., J. Electrochem. Soc. 135, 2669 (1988).
- 4) M. Medarde et al., Phys. Rev. Lett. 110, 266401 (2013).
- 5) M. C. Blesa et al., Solid State Ionics ${\bf 126},\, 81$ (1999).
- 6) T. Ichida et al., J. Phys. Soc. Jpn. **29**, 795 (1970).
- 7) T. McQueen et al., Phys. Rev. B 76, 024420 (2007).
- 8) J. Sugiyama et al., Phys. Rev. B 82, 224412 (2010).

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